

304(3): Calculation of Electric Dipole Transition Moments.

It has been shown in previous paper that the Evans Morris shifts are given by:

$$\left(\frac{\omega}{\omega_0}\right)^3 \left(\frac{e^{y_0} - 1}{e^y - 1}\right) = \exp(-Al) \quad - (1)$$
$$= \exp(-A\sqrt{t})$$

where the integrated power absorption coefficient for a narrow band is given by:

$$A = \left(\frac{N}{V}\right) \frac{|\mu_{gi}|^2}{6\epsilon_0 \sqrt{\pi}} \quad - (2)$$

Here ω_0 is the incident frequency, ω the shifted frequency,

$$y_0 = \exp\left(\frac{h\omega_0}{kT}\right) \quad - (3)$$

$$y = \exp\left(\frac{h\omega}{kT}\right) \quad - (4)$$

l is the sample path length, v is the velocity of light in the sample and:

$$l = vt. \quad - (5)$$

Therefore t is the time:

$$t = \frac{l}{v} \quad - (6)$$

taken for a beam to traverse the sample length. This

a) can be measured w/ time resolved apparatus. Here are N molecules in a sample volume V , ϵ_0 is the vacuum permittivity and \hbar is the reduced Planck constant.

The perturbation Hamiltonian is:

$$H^{(1)}(t) = -\underline{\mu} \cdot \underline{E}(t) \quad (7)$$

where $\underline{\mu}$ is an electric dipole moment and $\underline{E}(t)$ is the electric component of the electromagnetic field. For a plane polarized beam in the z axis:

$$H^{(1)} = -\mu_z E(t) = -e r \cos \theta \cdot E(t) \quad (8)$$

The selection rule for an electric dipole transition are:

$$\Delta n \text{ unrestricted}, \Delta l = \pm 1, \Delta m_l = 0, \pm 1 \quad (9)$$

For example to calculate the transition dipole moment for a $2s$ to $2p_z$ transition using Slater orbitals:

$$\mu_z = \int \psi_{2p_z} \mu_z \psi_{2s} d\tau \quad (10)$$

Zero:

$$\psi_{2s} = \left(\frac{Z^*{}^5}{96\pi a_0^5} \right)^{1/2} r \exp\left(-\frac{Z^* r}{2a_0} \right) \quad (11)$$

$$\psi_{2p_z} = \left(\frac{Z^*{}^5}{32\pi a_0^5} \right)^{1/2} r \cos \theta \exp\left(-\frac{Z^* r}{2a_0} \right) \quad (12)$$

So:

$$\mu_z = -\left(\frac{e}{\sqrt{3}}\right) \left(\frac{Z^{*5}}{32\pi a_0^5}\right) \int_0^{2\pi} \int_0^\pi \cos^2 \theta \sin \theta d\theta \int_0^\infty r^5 \exp\left(-\frac{Z^* r}{a_0}\right) dr$$

$$= -\left(\frac{5}{\sqrt{3} Z^*}\right) e a_0 \quad - (13)$$

Calculation for the Hydrogenic Wavefunction

The calculation for the hydrogenic wavefunction proceeds by:

$$\mu_z = \int \psi_1 e r \cos \theta \psi_2 d\tau \quad - (14)$$

where

$$\mu = e r \cos \theta \quad - (15)$$

where ψ_1 and ψ_2 are different functions of the H atom with:

$$\Delta l = \pm 1, \Delta m_l = 0, \pm 1. \quad - (16)$$

Here:

$$\int f d\tau = \int_{\phi=0}^{2\pi} \int_{\theta=0}^{\pi} \int_{r=0}^{\infty} f r^2 \sin \theta dr d\theta d\phi \quad - (17)$$

so the transition dipole moment is:

4)

$$\mu_z = \int_{\phi=0}^{2\pi} \int_{\theta=0}^{\pi} \int_{r=0}^{\infty} \phi_1 \cos \theta \phi_2 r^2 \sin \theta dr d\theta d\phi \quad (18)$$

and these can be calculated only if ϕ_1 and ϕ_2 are wave functions. The two wave functions ϕ_1 and ϕ_2 must satisfy the selection rules (16).

As a check on the calculation, μ_z should be zero if ϕ_1 and ϕ_2 do not satisfy the selection rules.

In the usual calculation of the integrated power absorption coefficient A , it is assumed that v is the speed of light. However in photon mass theory:

$$E = h\nu = \gamma mc^2 \quad (19)$$

$$p = h\underline{\nu} = \gamma m \underline{v} \quad (20)$$

and

$$\gamma = \left(1 - \frac{v^2}{c^2}\right)^{-1/2} \quad (21)$$

where

In an absorbing sample the refractive index is frequency dependent and given by:

$$n(\omega) = \frac{c}{v} \quad (22)$$

As in the next note, photon mass theory can be worked out to solve theory.