

224(7) : Simple Refutation of the Heisenberg Uncertainty Principle (UFT 175)

The Heisenberg uncertainty principle can be stated as:

$$\langle [\hat{q}, \hat{p}] \rangle = i\hbar \quad - (1)$$

where \hat{q} and \hat{p} are operators of quantum mechanics and where \hbar is the reduced Planck constant:

$$\hbar = \frac{h}{2\pi} \quad - (2)$$

It was asserted erroneously by Heisenberg that eq. (1) is a principle of indeterminacy, so that \hat{q} and \hat{p} may be "unknowable". These are non-Bacaria concepts rejected immediately by Einstein, de Broglie, Schrodinger and Dirac, and later by many other leading physicists.

In UFT 175 the anti-commutator equation was inferred:

$$[\hat{x}^2, \hat{p}] \psi = 2i\hbar [\hat{x}, \hat{p}] \psi \quad - (3)$$

where ψ is the wave function, \hat{x} the position operator and \hat{p} the momentum operator. Eq. (3) was evaluated by direct calculation using the

2) exact solutions of the Schrodinger equation: the harmonic oscillator, particle on a ring, spherical harmonics and wave functions of the hydrogen atom.

The following were calculated:

$$[x, p_x], [x, p_x^2], [x^2, p_x], [x^2, p_x^2], \\ [x, p_x^3], [x^3, p_x], [x^3, p_x^2],$$

for each exact solution of the Schrodinger equation. It was found that for the harmonic oscillator:

$$\langle [x^2, p_x^2] \rangle = 0 \quad \text{--- (4)}$$

and also for some other exact solutions. In the Heisenberg-Copenhagen interpretation eq. (4) means that x^2 and p_x^2 are "knowable". However, from eq. (1), the Copenhagen interpretation asserts that x and p_x can be "absolutely unknowable", i.e. that if p_x is knowable completely, x is completely unknowable. However, if x is "completely unknowable", it becomes "completely knowable" again from eq. (4), reductio ad absurdum. The Heisenberg uncertainty principle is obviously absurd.